

A HYBRID MADM ANALYSIS IN EVALUATING PROCESS OF CHEMICAL WASTEWATER PURIFICATION REGARDING TO ADVANCE OXIDATION PROCESSES

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Abstract. Organic components, widely spread in water by industrial sources become emerging Contaminants in the water. Due deficient removal in conventional wastewater treatments, advanced treatment should be considered for industrial wastewaters removing. Advance oxidation processes (AOPs) can be used as an effective subset of chemical treatment for degradation of toxic from water in industrial processing plant. Two Multiple Criteria Decision Making (MCDM) methods based on hybrid SWARA-WASPAS are organized to investigate efficiency of different sort of AOPs: ozonation, fenton, electrochemical Oxidation, UV/Photo-catalysis, UV/H₂O₂. Based on MCDM, Fenton is the most possible AOPs for application in wastewater treatment areas. In this study, SWARA is deployed in evaluating criteria that were identified in literature review, and WASPAS is used for evaluating and ranking the methods. The main point of this study is based on real data.

Keywords: wastewater, wastewater purification, AOPs, MCDM, SWARA, WASPAS.

Introduction

Population growth and industrialization caused more pollution of hydrosphere with organic and inorganic compounds (Oppenländer 2003). Sanitation is one of the essential practices in developing infrastructures in each government while encompasses wastewater treatment in each region of a country to fight for pollution and control it. Regarding recent UNICEF reports of 780 million people without access to safe drinking water, improvement in sanitation and purification methods needs critical attention. In fact purification is done to remove chemicals, biological and physical contaminants. Water purification is a process to reuse and recycle wastewaters for a specific purpose such as industrial usage.

In the literature four main sources of wastewaters are (1) domestic sewage, (2) industrial wastewaters, (3)

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agricultural runoff and (4) storm water and urban runoff. Although agricultural and urban runoff wastewaters have less importance, however domestic and industrial wastewater treatments are becoming increasingly critical due to rapid growth of industrialization and mechanization. Therefore all of the chemical compounds, materials and energy which are discharged into the water can potentially influence hydrosphere, global health and environment. This discharge can finally lead to transaction of energies and materials within an ecosystem's hydrology and can be a strong source of pollutant (contaminant). Role of AOPs as an effective chemical treatment is to degrade toxics and minimize the pollutants from the water that is investigated in this section. Almost all of the industries as cosmetics, wood and food and textile industries which contain chemical nature and processes can potentially produce water pollution.



Different treatment technologies are used to remove contaminants as environmental disinfectants and wastewater purification methods like chemical, physical, and biological operations. Wastewater purification is a complex process that needs to combine physical-chemical methods. Particularly, elimination of organic contaminations such as phenols, solvents, dye, etc., from wastewater is hard. Physical and mechanical treatment including filtration and membrane technique leads to secondary wastewater and higher cost. None of these treatment methods is effective enough to produce water with acceptable levels of organic compounds (Leyva-Díaz et al. 2015). Also, organic substances are stable compounds to be decomposed by common biological treatment. Literatures show that biological methods employed for degradation of organic compounds are ineffective and combined chemical- biological methods are suggested (Mantzavinos, Psillakis 2004). For instance, the biological treatments cannot be effective to remove most of complex poly-aromatic dye in textile industry. Chemical processes cannot be neglected and are highly recommended in decision making. Chemical treatment of wastewater may also be applied where there is limited access to the biological treatment.

Recent developments in the domain of chemical water treatment have led to an improvement in oxidative degradation procedures to remove organic materials in wastewater by oxidation through catalytic and photochemical methods. They are generally referred to advanced oxidation processes (AOPs) which has a strong impact on simple sanitation infrastructure decision and economical feasible approach. AOPs as chemical processes are promising methods to treat the most organic compounds from polluted water (Sievers 2011). It is a powerful tool for degradation and mineralization of pollutants in water and wastewater treatment, including hydrogen peroxide, ozone or special catalysts such as titanium dioxide. The use of a strong oxidizing agent can result in a high degree of wastewater treatment, including the breakdown of chemical recalcitrant and toxic compounds.

The aim of the present paper is to investigate the efficiency of different AOPs in wastewater treatment plant for wastewater management, namely Ozonation, Fenton, electrochemical Oxidation, UV/Photo-catalysis, UV/ H_2O_2 , and various integrated processes, regarding (i) efficiency and capacity of each technology for mineralization organic materials loads, (ii) comparable low cost and high cost technologies related economic and operational feasibility (iii) important chemical indicators to rank the possibility of techniques for removing effluents' toxicity. In this way, policy and decision making processes related to these issues are complicated. Multiple Criteria Decision Making (MCDM) (Liu *et al.* 2016; Liu, Shi 2015; Khan, Samadder 2015; Liu, Jin 2012; Turskis *et al.* 2012; Kucas 2010) frameworks can be a good choice for making an appropriate decision. Significance and contribution of multi criteria decision tools was applied in wide range of real life applications (Liu *et al.* 2011, 2012). Two MCDM methods are deployed in this study for making sound decisions. At first, SWARA is deployed for evaluating and weighting the criteria of this research. Must important criteria are identified from literature review. At next step, WASPAS method is deployed for evaluating different considerable alternatives of AOPs.

1. Literature review

In the literature review section, the significance of AOPs and its important parameters which have direct effect on term of AOPs wastewater treatment are discussed in Table 1. Several researches have reported applying different sort of AOPs and comparing them with conventional chemical methods. The high efficiency of AOPs in degradation of chemical toxin and pollution from water has been mentioned.

Sharma *et al.* (2012) compared advance and conventional oxidation processes for removing Microcystins (MCs) from water. It has been reported that AOPs are more effective and friendly to the environment than conventional chemical processes to convert a great variety of organic compounds into CO_2 and mineral acids. Study showed rate of pollution degradation with AOPs is faster than chlorine and chlorine dioxide as conventional methods.

Ozonation technology as initial AOPs has been scrutinized. Gómez-Pacheco *et al.* (2011) evaluated Ozonation process and showed a reasonable decrease in total organic carbon (TOC) performance with degradation of tetracyclines (TCs) from water. The result showed a rapid mineralization of organic compound and direct relation between operation variables and pollution degradation.

Several parameters are included to investigate different sort of AOPs, which make it hard to approach a prospective strategic decision making. Canizares *et al.* (2009a) compared three different AOPs such as: ozonation, electrochemical oxidation and Fenton. They illustrated that the efficiencies of these three technologies were critically depended on concentration of pollutant. Also, the operation cost analysis showed Fenton oxidation process had lower expenses than two others methods. In contrast, the cost calculation for two important oxidation processes, ozonation and Fenton reaction in a laboratory and pilot scale have been presented. This investigation showed Fenton reagent treatment was more expensive when compared to ozonation that is due to additional chemical cost for adjusting pH value of Fenton process (Krichevskaya *et al.* 2011).

Chang *et al.* (2008) took into account chemical and economic indicators for ozonation experiment. They investigated pollution and organic carbon removing were pH- dependent and the results showed the ozonation could not remove the pollutants entirely. The results declare a short ozone contact time has a better cost savings. Also, Fenton process has been under total cost and mineralization processes analyzes (Carra *et al.* 2013). They evaluated cost reduction has a direct correlation with the level of mineralization that higher mineralization levels cause significant decrease in financial terms.

Tizaoui *et al.* (2011) reported photo catalytic as an available AOPs technique for removing wastewater pollution by comparison with UV/H_2O_2 and ozonation processes. The results of this study successfully showed photo catalytic process was the most effective way for degradation pollution from water. Also, several researches tried to analyze the efficiency of different catalysts such as TiO_2 and ZnO for removing chemicals from water. They evaluated the optimum amount of catalyst, value of pH and the electrical energy cost as important indicators for reducing pollution from water (Lizama *et al.* 2002).

Exploring different advance oxidation processes (ozonation and its relatives, Fenton and photo catalysis) have been performed with ultra violet or solar source as energy consumption indicators (Esplugas *et al.* 2002). In addition, the research showed the ratio of photocatalysis process was five times lower than UV/H_2O_2 while the ability of recycling of photocatalyst powder could decrease the total cost of this process.

Cabral da Silva *et al.* (2013) used electrochemical oxidation to produce fresh water. Through this method they analyzed different parameters such as pH, total organic carbon (TOC) and energy consumption and mentioned electrochemical treatment method could be a feasible pretreatment process for the petrochemical industry. Also, Canizares *et al.* (2009a, 2009b) compared electrochemical oxidation method with other oxidation process. They believed electrolyte salt as a chemical parameter can improve efficiency and decrease total amount of pollution.

Autin *et al.* (2013) used an annular reactor to compare photocatalyst and hydrogen peroxide processes. It has been figured out that the optimization of reactor design, catalyst particle properties, and efficiency of UV lamp source will ultimately reduce the energy consumption. Moreover, different approaches presented dye pollution degradation using photocatalyst such as titanium dioxide (TiO₂). Some advances of this kind of oxidation have been performed for degradation of organic pollutants in wastewater (Teh, Mohamed 2011).

Chen *et al.* (2011) studied water pollution treatment for reusing water with TiO_2 photocatalyst method. They evaluated various parameters such as light source and retention time and mentioned that the proposed system can decrease the operating and maintenance costs of wastewater treatment for agricultural and domestic use.

2. The model of research

Table 1 illustrates the references of parameters as the model of research that has been done for adjusting the environmental, economical and chemical evaluation which was reported by authors. Table 2 demonstrates all information

Table 1. The model of research

Parameters	Reference
Mineralization	Lizama <i>et al.</i> 2002; Esplugas <i>et al.</i> 2002; Chang <i>et al.</i> 2008; Canizares <i>et al.</i> 2009a, 2009b; Tizaoui <i>et al.</i> 2011; Gómez-Pacheco <i>et al.</i> 2011; Krichevskaya <i>et al.</i> 2011; Autin <i>et al.</i> 2013; Prieto-Rodriguez <i>et al.</i> 2013; Carra <i>et al.</i> 2013; Carra <i>et al.</i> 2013; Carra
Volume of wastewater treated	Lizama <i>et al.</i> 2002; Esplugas <i>et al.</i> 2002; Chang <i>et al.</i> 2008; Canizares <i>et al.</i> 2009a, 2009b; Tizaoui <i>et al.</i> 2011; Gómez-Pacheco <i>et al.</i> 2011; Krichevskaya <i>et al.</i> 2011; Autin <i>et al.</i> 2013; Prieto-Rodriguez <i>et al.</i> 2013; Carra <i>et al.</i> 2013; Carra <i>et al.</i> 2013; Carra
Capital costs	Canizares et al. 2009b; Carra et al. 2013
Operating costs	Canizares <i>et al.</i> 2009a; Krichevskaya <i>et al.</i> 2011; Carra <i>et al.</i> 2013; Prieto-Rodriguez <i>et al.</i> 2013; Cabral da Silva <i>et al.</i> 2013:
Energy consumption	Lizama <i>et al.</i> 2002; Canizares <i>et al.</i> 2009b; Krichevskaya <i>et al.</i> 2011; Cabral da Silva <i>et al.</i> 2013; Autin <i>et al.</i> 2013
pH levels	Lizama <i>et al.</i> 2002; Esplugas <i>et al.</i> 2002; Gómez-Pacheco <i>et al.</i> 2011; Chang <i>et al.</i> 2008; Canizares <i>et al.</i> 2009a; Tizaoui <i>et al.</i> 2011; Krichevskaya <i>et al.</i> 2011; Autin <i>et al.</i> 2013; Prieto-Rodriguez <i>et al.</i> 2013; Carra <i>et al.</i> 2013; Cabral da Silva <i>et al.</i> 2013
Total time for decreasing pollution	Lizama <i>et al.</i> 2002; Esplugas <i>et al.</i> 2002; Chang <i>et al.</i> 2008; Canizares <i>et al.</i> 2009a, 2009b; Tizaoui <i>et al.</i> 2011; Gómez-Pacheco <i>et al.</i> 2011; Autin <i>et al.</i> 2013; Prieto-Rodriguez <i>et al.</i> 2013; Carra <i>et al.</i> 2013; Krichevskaya <i>et al.</i> 2011; Cabral da Silva <i>et al.</i> 2013
Amount of chemical material used	Lizama <i>et al.</i> 2002; Chang <i>et al.</i> 2008; Canizares <i>et al.</i> 2009a; Krichevskaya <i>et al.</i> 2011; Tizaoui <i>et al.</i> 2011; Autin <i>et al.</i> 2013; Prieto-Rodriguez <i>et al.</i> 2013
Pollution concentration	Krichevskaya <i>et al.</i> 2011; Cabral da Silva <i>et al.</i> 2013; Prieto-Rodriguez <i>et al.</i> 2013; Carra <i>et al.</i> 2013; Autin <i>et al.</i> 2013

about criteria and sub-criteria that are used in different aspects of decision making. These criteria are organized based on environmental, economic and chemical factors for five AOPs: Ozonation, Fenton, UV/Photo-catalysis, Electrochemical Oxidation and UV/H2O2. Environmental factors represent the capacity of each technology for elimination of organic pollution and their capability for mineralization organic materials to mineral materials (Saeed, Sun 2012). Economic study is prepared to compare low cost and high cost technologies related the pollutant treatment. These indicators include capital costs, operating costs and energy consumption. Capital costs involve expenses incurred for fixed equipment, installation and organic matter mineralization for an industrial-scale plant. Operating costs consist of treatment plant and all machinery that are required to operate a process. Furthermore chemical process of waste water treatment is a process that consumes electrical energy. This can be an important factor to determine operation costs. The volumetric electrical energy consumption per order (EE/O) and the electrical energy consumption cost are determined according to rated power (kW) of the AOPs system and radiation time (h) per volume (m³) of waste water that is treated (Kim *et al.* 2012). The third section of Table 2 is related to important chemical indicators such as total time for decreasing pollution, pH levels, and average amount of chemical materials that were used for treatment. The amounts of pollutions were studied in their basic amount in each treatment method. In this study the importance of pH level that has to be adjusted for different AOPs treatment has been considered. The pH level is important factor that can cause an increase in amount of pollution. Finally, the data in each column for each treatment method have been calculated in average such as the years of practice with the information extracted from references as mentioned in the Table 2.

Water treatment based AOPs acts in a way to reduce costs and increase process efficiency. There are multiple data that are reported in different articles and build a challenging situation for selecting suitable parameters. In this way the best parameters have been found within various articles.

Table 2. Environmental, economical and chemical factors for five AG	OPs
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				Ozonation	Fenton	Electro- chemical Oxidation	UV Photo- catalysis	UV H ₂ O ₂
Unit/ Para- meters/Ref		The treatment method		Gómez- Pacheco <i>et al.</i> 2011; Chang <i>et al.</i> 2008; Canizares <i>et al.</i> 2009a; Tizaoui <i>et al.</i> 2011; Krichevskaya <i>et al.</i> 2011	Canizares et al.2009; Prieto- Rodriguez et al. 2013; Carra et al. 2013; Krichevskaya et al. 2011; Canizares et al. 2009	Canizares <i>et al.</i> 2009a; Cabral da Silva <i>et al.</i> 2013; Canizares <i>et al.</i> 2009b	Tizaoui et al. 2011; Lizama et al. 2002; Autin et al. 2013; Prieto- Rodriguez et al. 2013	Tizaoui et al. 2011; Autin et al. 2013; Esplugas et al. 2002
Environ	%	Mineralization (C ₁₋₁)	Max	52.85	64.25	88.90	85	87.50
Environ- mental (C_1)	Litter	Volume of wastewater treated (C_{1-2})	Max	1.40	17.20	1.05	12.3	5.10
	€.m ⁻²	Capital costs (C ₂₋₁)	Min	88639	8568.50	50210	70400	10800
Economic	€.m ⁻³	Operating costs (C ₂₋₂)	Min	546.59	6.20	12.19	27.30	50.50
Unit/ Para- meters/Ref	KW h m ⁻³	Energy consumption $(C_{2,3})$ Min		2070	1	219.28	31.75	5.4
Unit/ Para- meters/Ref	-	pH levels (C ₃₋₁)	_	7	2.9	7.40	7.40	6.30
Chaminal	min	Total time for decreasing pollution (C_{3-2})	Min	82.93	50	150	153.75	33.75
$(C_{3)}$	mg/L	Amount of chemical material used (C_{3-3})	Min	54.41	1017.21	5025	434	696
	Mol/Lit	Pollution concentration (C_{3-4})	Max	1.23×10 ⁻³	5.2×10 ⁻⁴	4.79×10 ⁻³	2.87×10 ⁻³	5.67×10 ⁻⁵
Average Year of practice (C_4)	Year	-	Max	2010	2011	2010	2010	2009

3. Methodology

The general framework of this research is organized based on MCDM concept. The methodology of this research is established based on two different MCDM methods as a hybrid method. SWARA and WASPAS have been used together first time in a research in 2013. Hashemkhani Zolfani *et al.* (2013a) applied this hybrid method in a research for decision making in the business issue. Moreover in this short period of time some other researches are organized based on this new hybrid methodology. All of other researches based on SWARA-WASPAS are listed below:

Vafaeipour *et al.* (2014) in assessment of regions priority for implementation of solar plants; Bitarafan *et al.* (2014) in evaluation of real-time intelligent sensors for structural health monitoring of bridges.

3.1. Step-wise Weight Assessment Ratio Analysis (SWARA)

This method is focused on criteria and not alternatives. SWARA method is presented by Keršulienė *et al.* (2010) for evaluating criteria with a new perspective. There are some other methods for evaluating criteria like AHP, ANP and FARE. All these methods are organized based on pairwise comparison and related to direct relations of criteria. During these years after 2010, applications of SWARA method has widened. Also the main logic of this method has developed in some researches like: (Hashemkhani Zolfani, Saparauskas 2013) and (Hashemkhani Zolfani, Bahrami 2014) in general dimensions of method. Both previous researches believed that SWARA method is so suitable for policy making and decision making in top level. In fact each issue should be solved based on special circumstances eventually this topic should be considered. The experts are the main source of competency in SWARA methodology.

In comparison to other methods like AHP and ANP; SWARA has totally different characteristics. At the first step experts should make decision about priority of each issue. The priority should be based on needs and situation. SWARA can be deployed easily in abnormal or complicated situation. Experts are freer to make decision. At the next step experts should express their opinion about values of criteria. S_j is a symbol for this issue and each expert should express his/her opinion as a percentage. Percentage's units in all previous researches were based on 5% intervals.

Application of this method has increased in recent years. This method was applied in different methodological applications. In some researches SWARA is deployed singly but also this method developed in some new hybrid MCDM methodologies like SWARA-VIKOR, SWARA-COPRAS, SWARA-COPRAS-G and SWARA-WASPAS. All developments of decision making models based on SWARA method up to now are listed in Table 3.

Table 3. Methodological development of SWARA method

	Topic	Methodology	Reference
1	Design of products	SWARA	Hashemkhani Zolfani <i>et al.</i> (2013a)
2	Investigating on the success factors of online games based on explorer	SWARA	Hashemkhani Zolfani et al. (2013b)
3	Prioritizing Sustainability Assessment Indicators of Energy System	SWARA	Hashemkhani Zolfani, Saparauskas (2013)
4	Supplier selection in agile environment	SWARA-VIKOR	Alimardani et al. (2013)
5	Investment prioritizing investment in high tech industries	SWARA-COPRAS	Hashemkhani Zolfani, Bahrami (2014)
6	Decision making on business issues with foresight perspective	SWARA-WASPAS	Hashemkhani Zolfani <i>et al.</i> (2013c)
7	Evaluation of real-time intelligent sensors for structural health monitoring of bridges	SWARA-WASPAS	Bitarafan <i>et al.</i> (2014)
8	Assessment of regions priority for implementation of solar plants	SWARA-WASPAS	Vafaeipour et al. (2014)
9	Personnel selection	Game Theory-SWARA	Hashemkhani Zolfani, Seyed Agha Banihashemi (2014)
10	External wall insulation	SWARA-TODIM	Ruzgys et al. (2014)
11	Selection of a Packaging Design	SWARA	Stanujkic et al. (2015)
12	Glasshouse locating	SWARA-COPRAS	Haghnazar Kouchaksaraei et al. (2015)
13	Technology Foresight About R&D Projects Selection	SWARA	Hashemkhani Zolfani et al. (2015)
14	Planning the priority of high tech industries	SWARA-WASPAS	Ghorshi Nezhad et al. (2015)

The original procedure to the criteria weights determination using SWARA can be expressed as follows:

Step 1 – All criteria should be sorted based on expert ideas (Zavadskas, Vilutienė 2006).

Step 2 – From the second criterion, the comparative importance of the average value s_j should be determined as follows: the relative importance of the criterion j in relation to the previous (j-1) criterion (Stanujkic *et al.* 2015).

Step 3 – Determine the coefficient k_i :

$$k_{j} = \begin{cases} 1 & j = 1 \\ s_{j} + 1 & j > 1 . \end{cases}$$
(1)

Step 4 – Determine the recalculated weight w_i :

$$w_{j} = \begin{cases} 1 & j = 1 \\ \frac{x_{j-1}}{k_{j}} & j > 1 \end{cases}$$
(2)

Step 5 – The final step in the calculation of criteria weights:

$$q_j = \frac{w_j}{\sum_{k=1}^n w_j},\tag{3}$$

where q_j denotes the relative weight of the criterion *j*. The calculating part of SWARA method is from determination of criteria weights till the end.

3.2. Weighted Aggregates Sum Product Assessment (WASPAS)

WASPAS method is a new MCDM method based on Weighted Sum Model (WSM) and Weighted Product Model (WPM). Zavadskas *et al.* (2012) illustrated that this aggregation makes WASPAS better in comparison with WSM and WPM. WASPAS calculation is based on these steps:

3.2.1. Normalized decision making matrix based on:

$$\overline{x}_{ij} = \frac{x_{ij}}{\underset{i}{opt x_{ij}}}, \text{ where } i = \overline{1, m}; \ j = \overline{1, n}.$$
(4)

If opt value is max

$$\overline{x_{ij}} = \frac{opt x_{ij}}{x_{ij}}, \text{ where } i = \overline{1, m}; \ j = \overline{1, n}.$$
(5)

If opt value is min.

3.2.2. Calculating WASPAS weighted and normalized decision making matrix for summarizing part:

$$= - \overline{x_{ij,sum}} = \overline{x_{ij}q_j}, \text{ where } i = \overline{1,m}; \quad j = \overline{1,n} .$$
 (6)

3.2.3. Calculating WASPAS weighted and normalized decision making matrix for multiplication part:

$$= -q_j$$

 $x_{ij,mult} = x_{ij}^{q_j}$, where $i = \overline{1, m}; j = \overline{1, n}$. (7)

3.2.4. Final calculating for evaluating and prioritizing alternatives based on:

$$WPS_i = 0.5 \sum_{j=1}^{n} x_{ij}^{m} + 0.5 \prod_{j=1}^{n} x_{ij}^{m}$$
, where $i = \overline{1, m}$; $j = \overline{1, n}$. (8)

All the researches based on the WASPAS method up to now are described in several references as below:

Zavadskas *et al.* (2012) in developing WASPAS as a new methodology; Hashemkhani Zolfani *et al.* (2013c) in decision making on business issues with foresight perspective; Dejus and Antucheviciene (2013) in assessment of health and safety solutions at a construction site; Šiožinytė and Antuchevicienė (2013) in solving the problems of daylighting and tradition continuity in a reconstructed vernacular building, Hashemkhani Zolfani *et al.* (2016a) in presenting Prospective Multiple Attribute Decision Making (PMADM) and Hashemkhani Zolfani *et al.* (2016b) in presenting Multiple Attribute Decision Making (MADM) based scenarios.

4. Experts' information

In the process of this research, eight experts were participating. They were from different field of expertise such as: Chemistry, Environmental engineering, Chemical engineering, Electrical engineering, Industrial engineering, Mechanical engineering and Civil engineering. All the general information about them is shown in Table 4.

5. Findings

As mentioned before, this research is organized based on two MCDM methods as a hybrid method. This section is includes two sections. The first part is results of SWARA method and the second part is related to the results of WASPAS method. These two sections are prepared based on experts' evaluation.

5.1. SWARA's result

In this section priority of criteria and sub-criteria is shown. All information about priority and weights of criteria and sub-criteria is illustrated in Tables 5–8. S_j's are calculated based on average of experts' evaluations. Priority of all criteria is defined again based on experts' evaluation similar to SWARA's methodology.

The weightage of main criteria and their priority are shown in Table 5. Environmental criteria were the most important in general.

	1	2	3	4	5	6	7	8
Gender	female	male	male	male	female	female	male	male
Field of Study	Chemistry	Environmental engineering	Chemical engineering	Chemistry	Electrical engineering	Industrial Engineering	Mechanical engineering	Civil engineering
Education Background	PhD	Master	PhD	PhD	Master	PhD	Bachelor	Bachelor
Work experience	15	18	20	10	10	12	16	14

Table 4. Background information of experts

Table 5. Final results of SWARA method in weighting criteria

Criterion	The comparative importance of average value s_j	Coefficient $k_j = s_j + 1$	Recalculated weight $w_j = \frac{x_{j-1}}{k_j}$	Weight $q_j = \frac{w_j}{\sum w_j}$
C_1		1	1.0000	0.3305
C ₂	0.2188	1.2188	0.8205	0.2712
C3	0.2313	1.2313	0.6664	0.2203
C ₄	0.2375	1.2375	0.5385	0.1780

Table 6. Final results of SWARA method in weighting sub-criteria of environmental

Criterion	The comparative importance of average value s_j Coefficient $k_j = s_j + 1$		Recalculated weight $w_j = \frac{x_{j-1}}{k_j}$	Weight $q_j = \frac{w_j}{\sum w_j}$	Final weights
C ₁₋₁		1	1	0.5416	0.1790
C ₁₋₂	0.1813	1.1813	0.8465	0.4584	0.1515

Table 7. Final results of SWARA method in weighting sub-criteria of economic

Criterion	The comparative importance of average value s _j	Coefficient $k_j = s_j + 1$	Recalculated weight $w_j = \frac{x_{j-1}}{k_j}$	Weight $q_j = \frac{w_j}{\sum w_j}$	Final weights
C ₂₋₁		1.0000	1.0000	0.4001	0.1085
C ₂₋₂	0.2313	1.2313	0.8121	0.3249	0.0881
C ₂₋₃	0.1813	1.1813	0.6875	0.2750	0.0746

Table 8. Final results of SWARA method in weighting sub-criteria of chemical

Criterion	The comparative importance of average value s_j	Coefficient $k_j = s_j + 1$	Recalculated weight $w_{j} = \frac{x_{j-1}}{k_{j}}$	Weight $q_j = \frac{w_j}{\sum w_j}$	Final weights
C ₃₋₂		1	1.0000	0.3167	0.0698
C ₃₋₃	0.1938	1.1938	0.8377	0.2653	0.0584
C ₃₋₄	0.1750	1.175	0.7129	0.2258	0.0497
C ₃₋₁	0.1750	1.175	0.6067	0.1922	0.0423

The final weightage of each sub-criteria is calculated based on its criteria. Results of SWARA method are applied in next step.

5.2. WASPAS's results

Evaluating of four different alternatives is organized in this section based on WASPAS methodology. All alternatives

evaluated based on criteria and sub-criteria. All the procedures of this methodology are shown in Tables 9–13.

Different sort of AOPs were ranked base on SWARA-WASPAS results. Five potential advance oxidation processes in this paper have been introduced as alternatives for wastewater purification plant. Priority of alternatives is shown in Table 13 based on calculating formula 4 and

	C ₁		C ₂			C3				C_4
	C ₁₋₁	C ₁₋₂	C ₂₋₁	C ₂₋₂	C ₂₋₃	C ₃₋₁	C ₃₋₂	C ₃₋₃	C ₃₋₄	C_4
Weight	0.1790	0.1515	0.1085	0.0881	0.0746	0.0423	0.0698	0.0584	0.0497	0.1780
	Max	Max	Min	Min	Min	7 (Opt)	Min	Min	Max	Max
Ozonation	52.85	1.40	88639	546.59	2070	7	82.93	54.41	1.23×10 ⁻³	2010
Fenton	64.25	17.20	8568.50	6.20	1	2.9	50	1017.21	5.2×10 ⁻⁴	2011
Electrochemical Oxidation	88.90	1.05	50 210	12.19	219.28	7.40	150	5025	4.79×10 ⁻³	2010
UV/ Photo- catalysis	85	12.3	70 400	27.30	31.75	7.40	153.75	434	2.87×10 ⁻³	2010
UV/H ₂ O ₂	87.50	5.10	10 800	50.50	5.4	6.30	33.75	696	5.67×10 ⁻⁵	2009

* All the criteria are real data. Also there is an exception and that is C_{3-1} . The optimum rate for this criterion is 7 and max or min isn't suitable for this criteria.

Table 10. WASPAS normalized decision making matrix

	C ₁		C ₂			C ₃				C ₄
	C ₁₋₁	C ₁₋₂	C ₂₋₁	C ₂₋₂	C ₂₋₃	C ₃₋₁	C ₃₋₂	C ₃₋₃	C ₃₋₄	C_4
Weight	0.1790	0.1515	0.1085	0.0881	0.0746	0.0423	0.0698	0.0584	0.0497	0.1780
	Max	Max	Min	Min	Min	7 (Opt)	Min	Min	Max	Max
Ozonation	0.5945	0.0814	0.0967	0.0113	0.0005	1.0000	0.4070	1.0000	0.2568	0.9995
Fenton	0.7227	1.0000	1.0000	1.0000	1.0000	2.4138	0.6750	0.0535	0.1086	1.0000
Electrochemical Oxidation	1.0000	0.0610	0.1707	0.5086	0.0046	0.9459	0.2250	0.0108	1.0000	0.9995
UV/ Photocatalysis	0.9561	0.7151	0.1217	0.2271	0.0315	0.9459	0.2195	0.1254	0.5992	0.9995
UV/H ₂ O ₂	0.9843	0.2965	0.7934	0.1228	0.1852	1.1111	1.0000	0.0782	0.0118	0.9990

This table is calculated based on formula 1.

Table 11. WASPAS weighted and normalized decision making matrix for summarizing part

	C ₁		C2			C3				C_4
	C ₁₋₁	C ₁₋₂	C ₂₋₁	C ₂₋₂	C ₂₋₃	C ₃₋₁	C ₃₋₂	C ₃₋₃	C ₃₋₄	C_4
Ozonation	0.1064	0.0123	0.0105	0.0010	0.0000	0.0423	0.0284	0.0584	0.0128	0.1779
Fenton	0.1294	0.1515	0.1085	0.0881	0.0746	0.1021	0.0471	0.0031	0.0054	0.1780
Electrochemical Oxidation	0.1790	0.0092	0.0185	0.0448	0.0003	0.0400	0.0157	0.0006	0.0497	0.1779
UV/ Photocatalysis	0.1711	0.1083	0.0132	0.0200	0.0023	0.0400	0.0153	0.0073	0.0298	0.1779
UV/H ₂ O ₂	0.1762	0.0449	0.0861	0.0108	0.0138	0.0470	0.0698	0.0046	0.0006	0.1778

This table is calculated based on formula 2.

	C C		C ₂			C ₃				C_4
	C ₁₋₁	C ₁₋₂	C ₂₋₁	C ₂₋₂	C ₂₋₃	C ₃₋₁	C ₃₋₂	C ₃₋₃	C ₃₋₄	C_4
Ozonation	0.9111	0.6838	0.7761	0.6739	0.5658	1.0000	0.9392	1.0000	0.9347	0.9999
Fenton	0.9435	1.0000	1.0000	1.0000	1.0000	1.0380	0.9729	0.8428	0.8955	1.0000
Electrochemical Oxidation	1.0000	0.6547	0.8254	0.9422	0.6689	0.9977	0.9011	0.7677	1.0000	0.9999
UV/ Photo- catalysis	0.9920	0.9505	0.7957	0.8776	0.7726	0.9977	0.8996	0.8858	0.9749	0.9999
UV/H ₂ O ₂	0.9972	0.8318	0.9752	0.8313	0.8818	1.0045	1.0000	0.8617	0.8021	0.9998

Table 12. WASPAS weighted and normalized decision making matrix for multiplication part

This table is calculated based on formula 3.

Table 13. The results of WASPAS

	$0.5\sum_{j=1}^{n} \overline{Xij}, sum$	$0.5\prod_{j=1}^{n} \stackrel{=}{\underset{X_{ij},mult}{\underset{x_{ij},mult}$	WSP _i	Ranking
Ozonation	0.2250	0.0809	0.3059	5
Fenton	0.4439	0.3596	0.8035	1
Electrochemical Oxidation	0.2679	0.1175	0.3855	4
UV/ Photocatalysis	0.2927	0.1971	0.4898	3
UV/H ₂ O ₂	0.3158	0.2058	0.5216	2

in final column shows ranking. After creating decision matrix and other steps for receiving the final ranking, according to Table 13 which shows the ultimate results of WASPAS methodology, Alternative 2 (Fenton) is the best reagent for a chemical treated section in industrial wastewater plant.

Conclusions

In the current era many techniques are proposed to treat industrial wastewater especially chemical treatment which has been developed to help the wastewater treatment systems. Selecting an appropriate chemical treatment method which can rely on less expenses and more environmental impacts can be interesting for decision makers in managing water pollution crises. Since Advance Oxidation Processes system (AOPs) is one of the leading chemical treatments method in degradation and mineralization of pollutants, this research has been accomplished based on it. This research followed five important sorts of AOPs: Ozonation, Fenton, electrochemical Oxidation, UV/Photo-catalysis, and UV/H₂O₂ which are compared by MCDM methods. SWARA is applied for evaluating and weighting the criteria and WASPAS method is deployed for evaluating distinctive alternatives of the AOPs.

Different case studies have been chosen to check the applicability of influential factors such as ability of each method to mineralize the pollution, capital and operating costs, pH level as a significant factor of industry outlet and the amount of chemicals that are used in each technology. Eight experts from distinct scientific fields participated in this research and integrated SWARA-WASPAS decision model showed that the Fenton process is the best option in AOPs area for removing many hazardous and organic pollutants regarding environmental, economical and chemical factors. The results of this investigation can be used for decision making in real cases of future researches. Also, governments can use this methodology for basic wastewater treatment process.

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